Natural Product Synthesis

Total Synthesis of (—)-Colombiasin A and (—)-Elisapterosin B**

David C. Harrowven,* David D. Pascoe, Daniela Demurtas, and Heather O. Bourne

Colombiasin A (1) and elisapterosin B (2) are recent additions to the family of diterpenes from the gorgonian octacoral *Pseudopterogorgia elisabethae*.^[1,2] Discovered by Rodriguez et al., their unusual molecular architecture soon attracted attention from the natural products community.^[3] Initially, colombiasin A proved the more popular target,^[4] with total syntheses being reported firstly by Nicolaou et al.^[5] and then by Kim and Rychnovsky.^[6] The latter report also describes the first total synthesis of (–)-elisapterosin B, a compound that exhibits strong antiplasmodial activity against *Plasmodium falciparum*, the parasite responsible for the most severe forms of malaria.^[3] A synthesis of (+)-elisapterosin B by Rawal et al. followed soon after.^[7] Herein we describe total syntheses of both (–)-colombiasin A (1) and (–)-elisapterosin B (2)

in which a Moore rearrangement^[8] of vinylcyclobutene **14** is used to set up intramolecular $[4+2]^{[4-6]}$ and $[5+2]^{[7,9]}$ cycloaddition reactions leading to the target compounds.

Our synthesis of **14** (Scheme 1) began with (–)-dihydrocarvone (**3**), which was reduced to alcohol **4** using LiAlH₄. Hydroboration with ((-)-ipc)₂BH, ^[10] followed by oxidative work-up gave diols **5** as a 5:2 mixture of diastereomers. These were separated by a combination of column chromatography and fractional crystallization. Sequential monotosylation to **6**, cyanide displacement to nitrile **7**, and DIBAL-H reduction to aldehyde **8** then allowed us to introduce the diene function by means of a Julia reaction with **9**.^[11] Higher yields were obtained using the Kocienski modification, ^[12] these conditions giving **10** as a 3:1 mixture of Z and E isomers in 79 % yield. Swern oxidation followed, and the resulting mixture of

[*] Dr. D. C. Harrowven, D. D. Pascoe, D. Demurtas, Dr. H. O. Bourne School of Chemistry University of Southampton

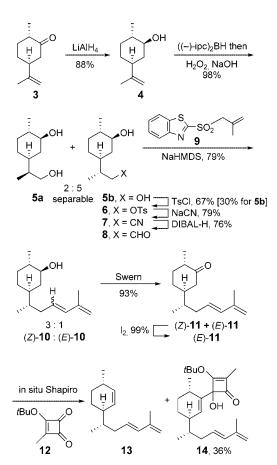
Highfield, Southampton, SO171BJ (UK)

Fax: (+44) 2380-596-805 E-mail: dch2@soton.ac.uk

[**] Financial support from EPSRC, Pfizer, and the University of Southampton is gratefully acknowledged.



Supporting information for this article is available on the WWW under $\frac{1}{\sqrt{w}}$ under



Scheme 1. Synthesis of intermediate **14**; ipc = 2,6,6-trimethylbicyclo[3.1.1]hept-3-yl, NaHMDS = sodium hexamethyldisilazane, Ts = p-toluenesulfonyl, DIBAL-H = diisobutylaluminum hydride.

Z and E dienones 11 was equilibrated to (E)-11 in near quantitative yield with catalytic iodine.

The next step, a Shapiro reaction between dienone (E)-11 and squarate 12,[13] proved troublesome.[14] Though the tosylhydrazone of (E)-11 was readily formed and isolated, it gave triene 13 as the major product on treatment with butyllithium and addition of squarate 12. By contrast, all attempts to prepare and isolate the corresponding trisylhydrazone (trisyl = triisopropylbenzenesulfonyl), either directly or indirectly, [15] met with failure. Monitoring the reaction by NMR spectroscopy showed that the formation of the trisylhydrazone was facile at room temperature in CDCl₃, and that it subsequently decomposed on prolonged standing. Consequently, an in situ variant of the Shapiro reaction was developed (Scheme 1). Trisylhydrazine and (E)-11 in THF were first stirred at ambient temperature for 2 h before the reaction mixture was cooled to -78°C. Four equivalents of *n*BuLi were then added and the temperature raised to -20 °C. At this juncture squarate 12 was added, giving vinylcyclobutene 14 in 36% yield.

The stage was now set for a Moore rearrangement of **14** to hydroquinone **16**,^[8] a reaction that proceeded smoothly at 110 °C in THF under microwave irradiation (Scheme 2). After cooling to ambient temperature and stirring in air, quinone **18** was isolated in a satisfying 80 % yield. Heating a toluene solution of **18** in the dark at 150 °C, either conven-

Zuschriften

Scheme 2. Total syntheses of (-)-colombiasin A (1) and (-)-elisapterosin B (2) starting with 14.

tionally or in the microwave oven, induced an intramolecular Diels–Alder cycloaddition to (–)-colombiasin A *tert*-butyl ether 17. [4-6] Attempts to effect removal of the protective group with TiCl₄ succeeded in that task, [16] but also led to Markovnikov addition of HCl across the sensitive C10–C11 double bond. [5] Deprotection with diethyl ether–trifluoroborane however, proceeded cleanly to complete a total synthesis of (–)-colombiasin A (1). Notably, exposing quinone 18 to diethyl ether–trifluoroborane induced both deprotection of the *tert*-butyl ether and an intramolecular [5+2] cycloaddition to give (–)-elisapterosin B (2); [6] our synthetic samples of 1 and 2 exhibit physical and spectral characteristics identical to those reported for the natural products. [1,2]

In conclusion, stereocontrolled syntheses of (-)-colombiasin A (1) and (-)-elisapterosin B (2) have been achieved, in twelve and eleven steps respectively from (-)-dihydrocarvone (3). The problematic C7 stereocenter was established by hydroboration with ((-)-ipc)₂BH, and the use of a *tert*-butyl protective group ensured that deprotection could be accomplished in high yield without disruption of the sensitive C10–C11 double bond.^[5] A distinctive feature of our approach is the use of a Moore rearrangement to set up intramolecular [4+2] and [5+2] cycloaddition reactions, leading to (-)-colombiasin A (1) and (-)-elisapterosin B (2) respectively. Studies are underway to further develop and improve the in situ Shapiro reaction and to apply the "reagent-free" rearrangement sequence in syntheses of related natural products such as elisapterosins A and D.^[2,3,17]

Received: October 11, 2004

Keywords: cycloaddition · diterpenes · marine natural products · rearrangement · total synthesis

- [3] a) A. D. Rodríguez, Y.-P. Shi, *Tetrahedron* 2000, 56, 9015 9023;
 b) A. D. Rodríguez, C. Ramírez, *J. Nat. Prod.* 2001, 2, 507 510.
- [4] a) D. C. Harrowven, M. J. Tyte, *Tetrahedron Lett.* 2001, 42, 8709 –
 8711; b) J. H. Chaplin, A. J. Edwards, B. L. Flynn, *Org. Biomol. Chem.* 2003, 1, 1842 1844.
- [5] a) K. C. Nicolaou, G. Vassilikogiannakis, W. Mägerlein, R. Kranich, Angew. Chem. 2001, 113, 2543–2547; Angew. Chem. Int. Ed. 2001, 40, 2482–2486; b) K. C. Nicolaou, G. Vassilikogiannakis, W. Mägerlein, R. Kranich, Chem. Eur. J. 2001, 7, 5359–5371.
- [6] A. I. Kim, S. D. Rychnovsky, Angew. Chem. 2003, 115, 1305–1308; Angew. Chem. Int. Ed. 2003, 42, 1267–1270.
- [7] N. Waizumi, A. R. Stankovic, V. H. Rawal, J. Am. Chem. Soc. 2003, 125, 13022 – 13023.
- [8] a) J. O. Karlsson, N. V. Nguyen, L. D. Foland, H. W. Moore, J. Am. Chem. Soc. 1985, 107, 3392-3393; b) S. T. Perri, H. J. Dyke, H. W. Moore, J. Org. Chem. 1989, 54, 2032-2034; c) A. Enhsen, K. Karabelas, J. M. Heerding, H. W. Moore, J. Org. Chem. 1990, 55, 1177-1185.
- [9] a) P. Walls, J. Padilla, P. Joseph-Nathan, F. Giral, J. Romo, *Tetrahedron Lett.* 1965, 1577-1582; b) P. Joseph-Nathan, V. Mendoza, E. Garcia, *Tetrahedron* 1977, 33, 1573-1576; c) P. Joseph-Nathan, M. E. Garibay, R. L. Santillan, *J. Org. Chem.* 1987, 52, 759-763; d) T. A. Engler, C. M. Scheibe, R. Iyengar, *J. Org. Chem.* 1997, 62, 8274-8275.
- [10] a) H. C. Brown, P. V. Ramachandran, J. Organomet. Chem. 1995, 500, 1–19; b) D. S. Matteson, Synthesis 1986, 973–985.
- [11] a) J. B. Baudin, G. Hareau, S. A. Julia, O. Ruel, *Tetrahedron Lett.* 1991, 32, 1175–1178; b) J. B. Baudin, G. Hareau, S. A. Julia, O. Ruel, *Bull. Soc. Chim. Fr.* 1993, 130, 336–357; c) J. B. Baudin, G. Hareau, S. A. Julia, R. Lorne, O. Ruel, *Bull. Soc. Chim. Fr.* 1993, 130, 856–878.
- [12] a) N. D. Smith, P. J. Kocienski, S. D. A. Street, Synthesis 1996, 652-666; b) P. R. Blakemore, P. J. Kocienski, A. Morley, K. Muir, J. Chem. Soc. Perkin Trans. 1 1999, 955-968; c) R. Bellingham, K. Jarowicki, P. Kocienski, V. Martin, Synthesis 1996, 285-296.
- [13] M. W. Reed, D. J. Pollart, S. T. Perri, L. D. Foland, H. W. Moore, J. Org. Chem. 1988, 53, 2477 – 2482.
- [14] A. R. Chamberlin, S. H. Bloom, Org. React. 1990, 39, 1-83.
- [15] H. Berner, H. Vyplel, G. Schulz, G. Fischer, *Monatsh. Chem.* 1985, 116, 1165–1176.
- [16] R. H. Schlessinger, R. A. Nugent, J. Am. Chem. Soc. 1982, 104, 1116–1118.
- [17] G. Zanoni, M. Franzini, Angew. Chem. 2004, 116, 4942-4946; Angew. Chem. Int. Ed. 2004, 43, 4837-4841

^[1] A. D. Rodríguez, C. Ramírez, Org. Lett. 2000, 2, 507-510.

^[2] A. D. Rodríguez, C. Ramírez, I. I. Rodríguez, C. L. Barnes, J. Org. Chem. 2000, 65, 1390 – 1398.